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### **Key indicators**

Single-crystal X-ray study T = 183 K Mean  $\sigma$ (C–C) = 0.002 Å R factor = 0.038 wR factor = 0.096 Data-to-parameter ratio = 29.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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# (4S,1'R)-Diethyl 6-methyl-2-[(1'-phenylethylimino)methyl]-4-(2-thienyl)-1,4-dihydropyridine-3,5-dicarboxylate

The absolute configuration of the title compound,  $C_{25}H_{28}N_2O_4S$ , has been determined. The molecules are interconnected by weak  $C-H \cdots O$  hydrogen bonds. The 1,4dihydropyridine (1,4-DHP) ring adopts the usual shallow boat conformation. The thiophene ring is nearly planar.

# Comment

Substituted 4-aryl-1,4-DHPs (1,4-DHP is 1,4-dihydropyridine) are biologically active, showing a wide range of effects, and have found applications in pharmacology as they affect cardiovascular diseases (hypertension, angina pectoris and cardiac arrhythmias), serve as calcium channel blockers (Goldmann & Stoltefuss, 1991), and show neurotropic, antidiabetic, antiviral, antibacterial and anticancer effects (Klusa, 1995; Gorlitzer et al., 2000; Sobolev et al., 2004).



(4R.1'R)- oil in filtrate

It has been recognized that the absolute configuration at the C4 position of the 1,4-DHP nucleus is crucial in modulating the biological activity (Triggle et al., 1980; Fossheim et al., 1982). Indeed, pairs of enantiomers that differ in their substituents at the C3 and C5 positions of 1,4-DHP usually differ in their biological properties; sometimes they can act in opposite directions, e.g. as calcium antagonist-calcium agonist (Kongsamut et al., 1985). The different biological activities of 1,4-DHP enantiomers and diastereomers can cause problems in drug development (Strong, 1999).

Three general methods have been reported for the preparation of enantiomeric 4-aryl-1,4-DHPs: (i) the optical resolution of 1,4-DHP-monocarboxylic acids (Goldmann & Stoltefuss, 1991); (ii) chemoenzymatic hydrolysis of the alkyl ester groups at the C3 and C5 positions of substituted 1,4-DHPs (Sobolev et al., 2004); (iii) the enantioselective Hantzsch-type synthesis using a chiral auxiliary on the 1,4-DHP N atom or the ester group (Kosugi et al., 1994; Peri et al., 2000). In our case, the third method was applied (Marchalín et al., 2004) for the preparation of the title compound, (I).

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#### Figure 1

The molecular structure of (I), showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

The expected stereochemistry of atoms C4 and C22 was confirmed as S and R, respectively (Fig. 1). There are intraand intermolecular  $C-H \cdots O$  hydrogen bonds in the crystal structure (Table 2). The 1,4-DHP ring adopts a shallow boat conformation, with atoms C4 and N1 deviating by 0.413 (1) and 0.181 (1) Å, respectively, from the planar base of the boat [the deviations from the mean plane are 0.0005 (17), 0.0005 (17), 0.0004 (16) and 0.0004 (17) Å for C2, C6, C3 and C5, respectively].

The puckering of the 1,4-DHP ring at C4 and N1 affects the biological activity of this class of compounds to a great extent (Triggle et al., 1980; Fossheim et al., 1982); see pertinent torsion angles involving atoms C4 and N1 (Table 1). The planar thiophene ring [maximum deviation from the mean plane is 0.005 (1) Å for atom C42] is approximately perpendicular to the 1,4-DHP ring. The dihedral angle between the plane of the five-membered ring and the plane of the base of the boat (C2/C3/C5/C6) is 86.98 (5)° (Nardelli, 1995). The C2-C3-C31-O31 and C6-C5-C51-O51 torsion angles are 13.4 (2) and 8.5 (2) $^{\circ}$ , respectively, reflecting the synperiplanar (sp, sp) conformations of the carbonyl C=O bonds.

## **Experimental**

The title compound was prepared by the reaction of the racemic aldehyde and chiral (R)- $\alpha$ -phenylethylamine [(R)- $\alpha$ -PEA] in dry ethanol. The title compound precipitated as crystals while the diastereomer diethyl (4R,1'R)-6-methyl-2-[(1'-phenylethylimino)methyl]-4-(2-thienyl)-1,4-dihydropyridine-3,5-dicarboxylate remained in the filtrate. Full details of the synthetic procedure have been published by Marchalín et al. (2004). Yellow prism-like single crystals were prepared by recrystallization from an ethanol solution (1:1).

Crystal data

C25H28N2O4S Z = 2 $D_x = 1.243 \text{ Mg m}^{-3}$  $M_r = 452.55$ Monoclinic, P2 Mo  $K\alpha$  radiation a = 10.3023 (2) Å  $\mu = 0.17 \text{ mm}^{-3}$ b = 8.1925 (1) Å T = 183 (2) K c = 14.3279 (2) Å Prism, yellow  $0.60 \times 0.28 \times 0.10 \text{ mm}$  $\beta = 90.420 \ (1)^{\circ}$ V = 1209.27 (3) Å<sup>3</sup>

#### Data collection

Siemens SMART CCD diffractometer (i) scans Absorption correction: multi-scan (SADABS; Sheldrick, 2002)  $T_{\min} = 0.902, \ T_{\max} = 0.984$ 

# Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0718P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.038$	+ 0.1736P]
$wR(F^2) = 0.096$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.05	$(\Delta/\sigma)_{\rm max} = 0.002$
8488 reflections	$\Delta \rho_{\rm max} = 0.29 \text{ e } \text{\AA}^{-3}$
293 parameters	$\Delta \rho_{\rm min} = -0.25 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	Absolute structure: Flack (
	3670 Friedel pairs

#### ute structure: Flack (1983), 3679 Friedel pairs

21384 measured reflections

 $R_{\rm int} = 0.025$ 

 $\theta_{\rm max} = 32.9^{\circ}$ 

8488 independent reflections 7741 reflections with  $F^2 > 2\sigma(F^2)$ 

# Flack parameter: -0.02(5)

#### Table 1 Selected geometric parameters (Å, °).

N1-C6	1.3830 (14)	C21-N21	1.2730 (15)
N1-C2	1.3853 (14)	C41-S41	1.7220 (12)
C2-C3	1.3599 (15)	S41-C44	1.7098 (15)
C6-N1-C2	122.35 (9)	C41-C4-C5	110.95 (9)
C3-C2-C21	126.19 (10)	C21-N21-C22	115.87 (10)
C2-C3-C31	122.86 (10)	C24-C22-C23	111.18 (10)
C3-C4-C41	109.57 (9)	C44-S41-C41	92.51 (7)
C6-N1-C2-C3	-18.34 (16)	C2-N1-C6-C5	17.82 (16)
N1-C2-C3-C4	-9.11(16)	N1-C2-C21-N21	-8.78 (16)
C2-C3-C4-C41	-89.68 (12)	N21-C22-C24-C25	45.16 (15)
C2-C3-C4-C5	32.53 (14)	C41-S41-C44-C43	0.23 (13)
C3-C4-C5-C6	-33.13 (14)	C5-C51-O52-C52	179.05 (11)

Table 2			
Hydrogen-bond	geometry	(Å,	°).

$\begin{array}{c ccccccccccccccccccccccccccccccccccc$					
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$D - H \cdots A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
	$C52 - H52B \cdots O31^{i}$ $C22 - H22 \cdots O51^{ii}$ $C21 - H21 \cdots O31$ $C4 - H4 \cdots O32$ $C4 - H4 \cdots O52$ $C61 - H61C \cdots O51$	0.99 1.00 0.95 1.00 1.00 0.98	2.61 2.41 2.31 2.26 2.31 2.35	3.043 (2) 3.403 (2) 2.898 (2) 2.699 (1) 2.671 (1) 2.903 (2)	107 171 120 105 100 115

Symmetry codes: (i) x - 1, v, z; (ii) x + 1, v, z.

The majority of the H atoms were visible in a difference Fourier map, including the methyl H atoms. H atoms were placed in idealized positions (N-H = 0.88 Å, methyl C-H = 0.98 Å, methylene C-H = 0.99 Å, methine C-H = 1.00 Å and aromatic C-H = 0.95 Å)and refined as riding, with  $U_{iso}(H) = 1.2U_{eq}$  (parent atom).

Data collection: *SMART* (Siemens, 1995); cell refinement: *SAINT* (Siemens, 1995); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Bruker, 2001); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 2002); software used to prepare material for publication: *SHELXL97*.

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# References

Brandenburg, K. (2002). *DIAMOND*. Version 2.1e. Crystal Impact GbR, Bonn, Germany.

Bruker (2001). SHELXTL. Version 6.10. Bruker AXS Inc., Madison, Wisconsin, USA.

Flack, H. D. (1983). Acta Cryst. A39, 876-881.

- Fossheim, R., Svarteng, K., Mostad, A., Romming, C., Shefter, E. & Triggle, D. J. (1982). J. Med. Chem. 25, 126–131.
- Goldmann, S. & Stoltefuss, J. (1991). Angew. Chem. Int. Ed. Engl. 30, 1559– 1578.
- Gorlitzer, K., Kramer, C. & Boyle, C. (2000). Pharmazie, 55, 651-658.
- Klusa, V. (1995). Drugs of the Future, 20, 135-138.
- Kongsamut, S., Kamp, T. J., Miller, R. J. & Sanguinetti, M. C. (1985). Biochem. Biophys. Res. Commun. 130, 141–148.
- Kosugi, Y., Hori, M. & Nagasaka, T. (1994). Heterocycles, 39, 591-602.
- Marchalín, Š., Cvopová, K., Kríž, M., Baran, P., Oulyadi, H. & Daich, A. (2004). J. Org. Chem. 69, 4227–4237.
- Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
- Peri, R., Padmanabhan, S., Rutledge, A., Singh, S. & Triggle, D. J. (2000). J. Med. Chem. 43, 2906–2914.
- Sheldrick, G. M. (2002). SADABS. Version 2.03. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Siemens (1995). *SMART* and *SAINT*, Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sobolev, A., Franssen, M. C. R., Duburs, G. & de Groot, A. (2004). *Biocatal. Biotransform.* 22, 231–252.
- Strong, M. (1999). Food Drug Law J. 54, 463-487.
- Triggle, A. M., Shefter, E. & Triggle, D. J. (1980). J. Med. Chem. 23, 1442-1445.